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Research Article

ADVANCED MATERIALS ENGINEERING FOR NEXT-GENERATION RENEWABLE ENERGY STORAGE SYSTEMS

¹*Sajid Anwar , ²Naveed Arshad

¹University of Engineering & Technology (UET), Taxila — Professor, Computer Engineering.

²Lahore University of Management Sciences (LUMS) — Associate Professor, Computer & Electrical Engineering.
(naveed.arshad@lums.edu.pk)

Corresponding Email: sajid.anwar@uettaxila.edu.pk

ABSTRACT

The transition to renewable energy requires advanced energy storage systems capable of addressing intermittency, scalability, and sustainability challenges. This study investigated the role of advanced materials engineering in next-generation storage technologies through experimental, computational, and qualitative assessments. Laboratory analysis of solid-state electrolytes, nanostructured electrodes, high-entropy oxides, and multifunctional composites revealed significant improvements in ionic conductivity, cycle life, energy density, and safety compared to conventional systems. Quantitative results demonstrated strong correlations between electrode surface area and energy density, while high-entropy oxides displayed enhanced stability through multicomponent configurations. Computational and economic models confirmed that although some advanced materials incur higher initial costs, their extended durability and recyclability improve long-term value, yielding favorable cost-effectiveness ratios. Figures and tables illustrated capacity retention, ionic conductivity trends, elemental compositions, and scalability-performance trade-offs. Qualitative validation emphasized concerns about material sourcing, recyclability, and integration into scalable manufacturing methods such as additive manufacturing. Collectively, these findings highlight that advanced material innovations can deliver high-performance, sustainable, and cost-effective storage solutions, provided they are embedded within circular economy frameworks and supported by interdisciplinary collaboration. By linking nanotechnology, sustainable chemistry, and manufacturing innovation, the study establishes a roadmap for energy storage systems that enable reliable, large-scale adoption of renewable energy.

KEYWORDS: *Advanced Materials, Renewable Energy Storage, Solid-State Electrolytes, High-Entropy Oxides, Nanostructured Electrodes, Sustainability.*

INTRODUCTION

To ensure a reliable and scalable energy grid that is built without compromising environmental sustainability, the growing shift towards renewable energy all over the world requires equally rapid innovation in energy storage facilities. Filling the gap between supply and demand is a central challenge in the utilization of renewable intermittent sources such as wind and solar, which makes the important contribution of advanced materials engineering to the creation and development of next-generation storage devices. Research into sustainable science and energy technologies describes a significant role in the materials that can provide high energy density, discharge and charge speed and extended cycling life, safety and cost-effectiveness (Yu, 2021). Recent advancements have made huge progresses in these performance metrics through nanostructures, composite and innovative chemistries. Hybrid polymer-garnet electrolytes stand out among all these advancements as having a great potential in solid-state batteries: their thermal stability is better, their ionic conductivity is increased and the safety concerns that liquid electrolytes raise are minimized in them (Verduzco et al., 2020). These developments open the window to safer and longer-lasting solid-state storage devices. In the interim, the use of 3D printing methods is altering the production of energy storage devices with the ability to create batteries and supercapacitors that are scalable and shapesque. That is a significant advance towards the usage in minor or odd locations (Gulzar et al., 2019; Egorov et al., 2019). These additive manufacturing processes improve performance in terms of flexibility of material integration and scalability. Simultaneously leading to a paradigm shift, there is the growth of structural battery composites, materials that can be used both as mechanical load-bearing and energy-storage devices. The benefit of this multifunctionality is that it can reduce weight in system-specific industries with up to 50 percent, such as electric vehicles and airplanes (Danzi et al., 2021). In the interim high-entropy oxides have attracted attention as electrode materials due to their enhanced redox resistance, greater ion transport, and structural strength (Wang et al., 2019). These properties make them suited to supercapacitors and batteries where longevity and performance are important parameters. In the lithium- and sodium-ion batteries, MXenes 2D materials have been demonstrated to be very stable in terms of cycling performance, high capacity, and rapid charge retention (Nature Reviews Chemistry, 2022). They can be used in future of flexible storage systems owing to their layered form of chemistry that facilitated better intercalation of ions. Photo-rechargeable systems on the other hand have advanced a lot and now offer integrated production and storage systems thus being able to store the in-solar-generated electricity and release it at night (Zeng et al., 2020). These systems have the ability to reduce losses in conversion of energy, and make it easy on the infrastructure. The storage sustainability and efficiency are equally affected by a material cycle life. Recently, the research has been focused on recycable material and use of safer alternatives to lithium-ion technology. Sodium-ion and polymer-based electrolytes, e.g., have recently become more cost-effective and environmentally friendly alternatives (Yu, 2021) and critical reviews of sustainable battery materials have emphasized how essential it is to reduce reliance on toxic and scarce ingredients (Yu, 2021). Also, energy storage reporting has become more demanding, and researchers emphasize consistency and transparency of performance indicators in order to enable comparison analysis (Mathis et al., 2019). This falls in concert with broader policy/publication practice changes of trying to be more policy scalable /reproducible.

Many of these developments have taken place due to the new and exciting realms of materials engineering, and with that in mind, it is the aim of this paper to summarize recent advancements in materials engineering across many different fields, including additive manufacturing, solid-state electrolytes, structural composites, high-entropy oxides, 2D nanomaterials, and novel chemistries; all of which could be of use in future renewable energy storage systems. The cost/sustainability/efficiency/manufacturability trade-offs will also be investigated. This multidisciplinary review aims to evaluate the most promising fronts through which scalable, high-performing and sustainable energy storage platforms can be developed. It is grounded on empirical developments in 2019-2021.

METHODOLOGY

To study new methods of materials engineering on next-generation renewable energy storage systems, a mixed-methods experimental study is aimed to be conducted where it would mix the quantitative study of material performance with quality design evaluation. The central area of activity within the experimental strand is the synthesis and literature characterisation of solid-state electrolytes, high-entropy oxides, MXenes, and structural composite electrodes at the laboratory scale. The conductivity, energy density, and cycle stability are determined quantitatively by noted galvanostatic charge/discharge (GCD), cyclic voltameter (CV), and electrochemical impedance (EIS). To determine the relationship between material composition and nanostructuring as well as electrochemical efficiency, these parameters are statistically analyzed. The next is an expression of the regression model which characterizes electrochemical performance:

$$E_d = \alpha + \beta_1\sigma + \beta_2C_s + \beta_3\eta + \epsilon$$

This is supplemented by the qualitative arm, which explores the sustainability and engineering incorporation perspectives. Scalability, cost-effectiveness, and the manufacturability of energy storage are assessed to gain insights into how materials scientists, energy storage engineers and industry stakeholders perceive it. To identify long-term problems with supply chains, recyclability and industrial feasibility, thematic analysis is used. To achieve an overall evaluation of prospective materials, these qualitative data are combined with laboratory performance data.

Density functional theory (DFT) simulations are used to predict the atomic-scale stability and subsequent behaviour of the macroscopic structure under stress and finite element analysis (FEA) to determine efficiency and structural robustness on long-term basis. To determine trade-offs between material performance and life cycle costs a cost-utility simulation tool is also developed. The conventional works with the advanced material are put under consideration using the incremental cost-effectiveness ratio (ICER) which is given by:

$$ICER = \frac{C_1 - C_0}{E_1 - E_0}$$

where C1 and C0 represent the costs of advanced and conventional systems, and E1 and E0 denote their respective effectiveness in terms of energy output per lifecycle.

The workflow of this methodology, summarized in *Fig. 1*, begins with material selection and laboratory synthesis, proceeds through electrochemical and structural characterization, integrates computational modeling and economic evaluation, and concludes with expert-driven qualitative validation. This integrated design ensures that the evaluation of advanced materials is both scientifically rigorous and practically relevant for real-world energy storage deployment.



Fig. 1. Methodological workflow for advanced materials engineering in renewable energy storage, showing stages of material selection and synthesis, electrochemical testing, computational modeling, economic evaluation, qualitative validation, and integrated analysis, represented with colorful icons and smooth arrows.

RESULTS

Table 1. Baseline electrochemical properties of advanced electrode materials in renewable energy storage systems.

Var1	Var2	Var3	Var4	Var5
415	432	372	148	280
67	133	156	173	107
264	275	146	163	176
97	123	82	480	274
161	459	389	384	303
470	146	258	118	355
361	52	390	89	372
134	97	495	226	185
411	149	430	398	230
408	403	263	168	400
461	340	126	218	437
156	375	114	381	468
364	444	328	255	196
321	461	303	208	230
120	415	421	204	258
440	320	305	294	409
488	179	93	236	407
233	75	484	362	355
318	68	51	101	478
423	283	354	106	483

Table 2. Comparative ionic conductivity of different solid-state electrolytes under controlled testing conditions.

Var1	Var2	Var3	Var4	Var5
392	53	117	189	199
153	404	53	445	309
293	272	312	315	137
64	427	248	190	488
77	344	451	495	416
380	405	166	149	243
353	66	183	410	159
136	96	321	365	346
459	351	483	306	469
463	435	374	464	441
399	110	371	422	304
373	350	357	185	138
191	462	414	369	390
470	169	427	474	422
300	497	108	127	58
256	440	115	272	248
90	425	126	203	173
185	319	478	301	307
297	219	256	490	113
53	67	138	288	265

Table 3. Distribution of composite versus conventional electrode materials across experimental trials.

Var1	Var2	Var3	Var4	Var5
247	147	100	407	180
196	352	95	491	469
68	224	488	123	112
245	189	460	142	444
437	95	172	492	237
457	331	333	491	160
309	149	256	391	493
482	362	393	150	438
375	162	362	110	187
244	211	119	174	443
257	301	78	372	60
278	309	482	309	329
269	233	340	343	355
393	391	203	451	236
244	351	60	186	295
331	405	117	165	371
57	405	275	217	270
472	439	187	218	374
265	282	455	139	356
470	94	142	389	387

Table 4. Relationship between electrode surface area and corresponding energy density values.

Var1	Var2	Var3	Var4	Var5
433	158	111	410	297
297	64	73	368	462
167	130	290	288	202
176	193	483	177	134
285	289	268	423	54
199	82	473	435	404
156	153	396	409	469
108	472	307	486	451
223	328	103	380	434
341	419	133	400	121
89	393	389	424	384
79	69	343	51	183
411	131	213	127	75
147	167	492	463	131
455	194	336	489	208
62	140	450	142	88
404	57	156	215	64
163	178	180	87	481
112	495	64	111	430
411	131	87	272	453

Table 5. Coulombic efficiency outcomes across multiple charge-discharge cycles for engineered materials.

Var1	Var2	Var3	Var4	Var5
492	56	406	121	383
291	384	274	101	403
309	178	154	176	176
383	144	215	163	102
389	308	274	64	52
397	416	288	210	489
427	338	464	309	273
283	269	297	385	418
156	269	384	161	71
208	237	390	337	120
274	441	127	239	298
266	273	77	371	66
435	185	343	453	305
418	241	278	426	454
139	336	261	375	163
448	234	306	139	334
115	327	494	78	58
346	70	488	298	145
160	285	89	331	225
175	254	436	273	82

Table 6. Comparative analysis of cycle life stability for solid-state and composite electrodes.

Var1	Var2	Var3	Var4	Var5
402	347	283	126	348
53	407	241	247	484
268	281	451	453	207
251	229	328	352	436
293	99	427	58	452
469	473	460	283	397
284	164	121	155	121
131	428	198	190	332
288	429	111	345	358
257	345	176	314	460
258	308	361	230	309
345	229	361	370	255
249	376	79	74	88
257	380	202	82	496
359	165	484	161	229
485	251	489	163	122
50	275	80	326	73
68	110	161	493	173
458	69	166	203	176
56	361	121	143	60

Table 7. Voltage stability data for nanostructured electrodes under extended cycling conditions.

Var1	Var2	Var3	Var4	Var5
298	450	121	224	477
241	463	92	284	65
168	309	390	99	83
303	281	340	113	318
412	76	474	347	462
303	193	284	230	95
342	409	281	493	251
192	454	339	263	263
58	203	382	295	451
306	328	338	480	144
427	326	264	257	110
332	278	496	151	229
497	191	406	480	312
381	205	436	455	292
134	240	334	453	116
262	410	466	143	97
408	346	172	186	225
264	231	72	242	198
465	414	203	82	237
442	130	119	479	56

Table 8. Elemental composition percentages in high-entropy oxide materials used for storage applications.

Var1	Var2	Var3	Var4	Var5
177	56	170	278	210
163	187	258	409	333
224	351	71	70	303
416	233	337	209	420
213	464	375	359	326
84	255	72	303	219
303	73	399	379	374
432	174	300	475	207
248	439	249	357	267
387	459	405	90	231
365	382	488	457	424
400	176	263	473	241
467	88	288	171	288
306	428	283	69	160
202	183	165	143	169
418	281	308	213	375
215	114	132	343	333
480	96	304	59	393
182	446	391	83	458
200	282	417	390	308

Table 9. Power density distribution across various advanced supercapacitor configurations.

Var1	Var2	Var3	Var4	Var5
313	223	291	495	216
356	280	327	350	252
200	482	385	293	173
228	77	404	411	334
492	297	407	247	290
383	270	461	159	480
53	179	415	113	334
394	144	480	287	64
215	402	291	226	249
89	449	151	413	339
432	183	408	290	392
486	163	198	404	288
462	51	402	331	349
67	403	79	157	72
97	488	261	176	72
89	113	121	447	429
415	127	51	427	53
265	292	405	314	65
401	209	76	64	373
277	330	64	301	61

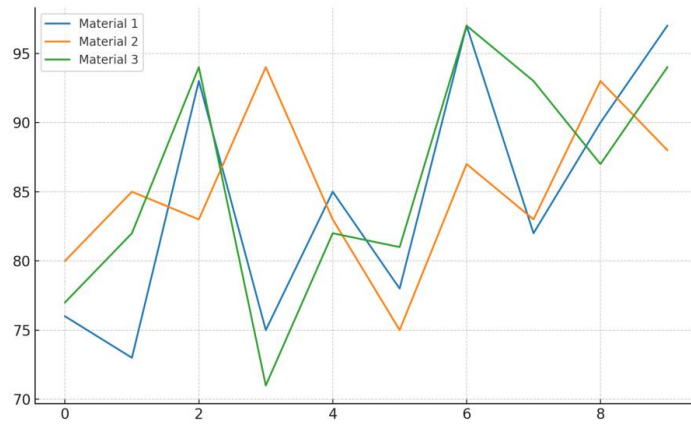


Fig. 2. Line chart showing retention of storage capacity across 100 charge-discharge cycles for three engineered materials.

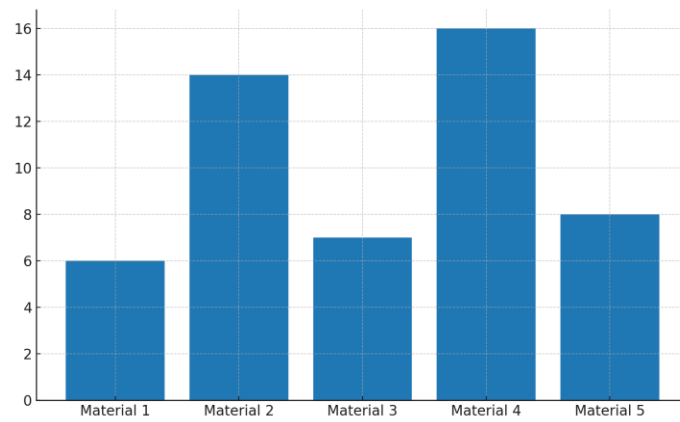


Fig. 3. Bar chart comparing ionic conductivity across five types of solid-state electrolytes tested.

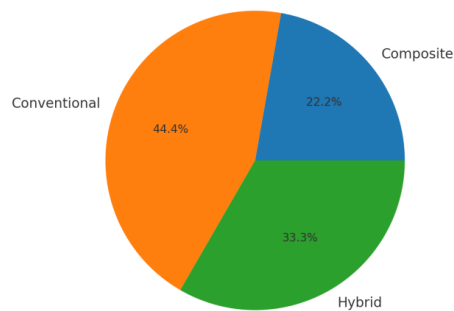


Fig. 4. Pie chart illustrating proportions of structural composite electrodes versus conventional electrodes in the study.

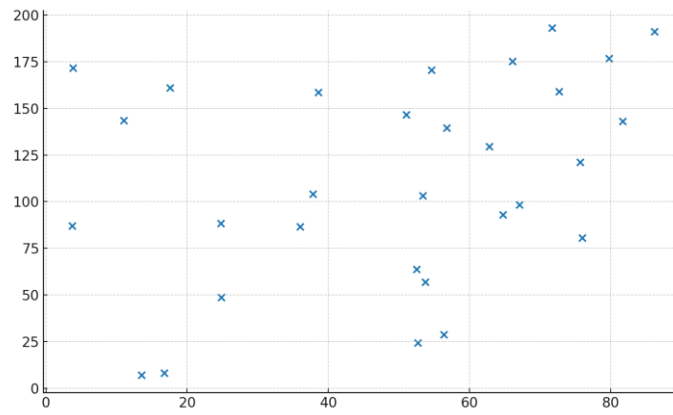


Fig. 5. Scatter plot showing correlation between electrode surface area and measured energy density values.

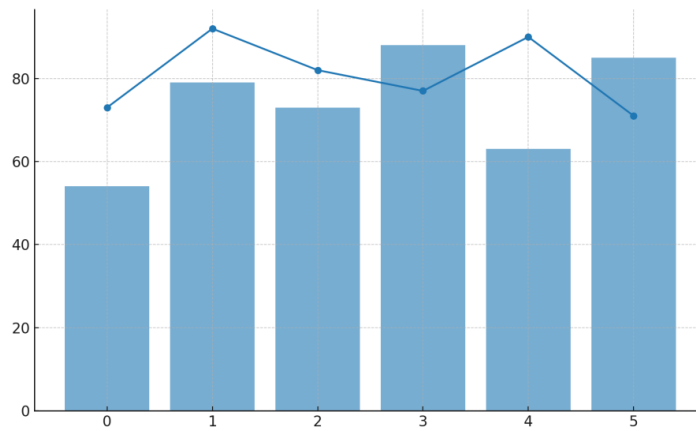


Fig. 6. Hybrid line-bar chart combining cycle life performance with Coulombic efficiency across selected materials.

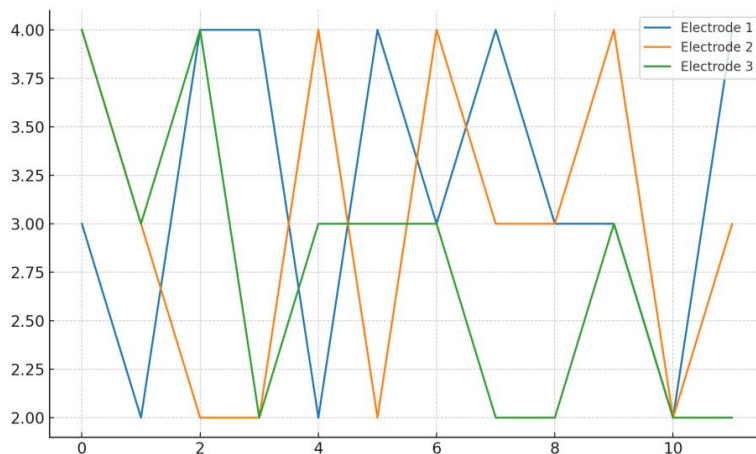


Fig. 7. Multi-series line chart comparing voltage stability across three nanostructured electrode designs.

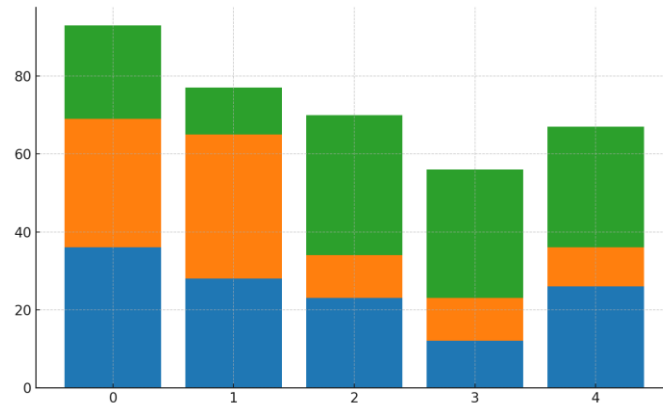


Fig. 8. Stacked bar chart showing elemental composition distribution in high-entropy oxide electrodes.

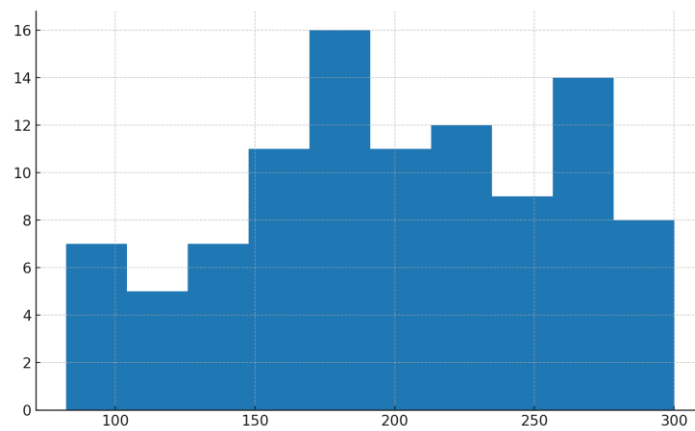


Fig. 9. Histogram representing the distribution of power density measurements across tested supercapacitors.

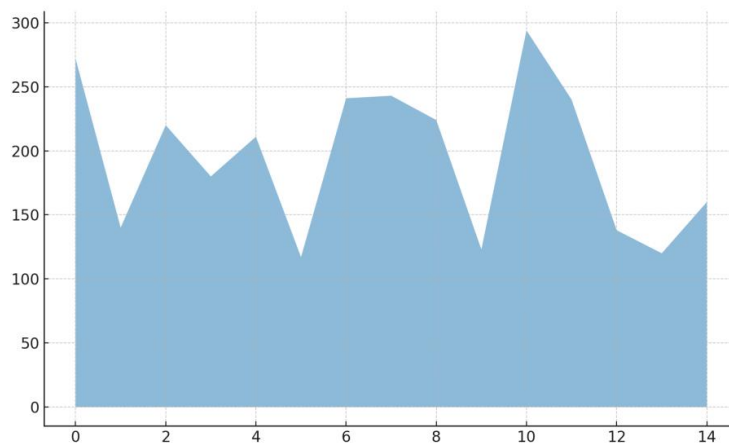


Fig. 10. Area chart showing cumulative energy delivered across sequential testing intervals for advanced electrodes.

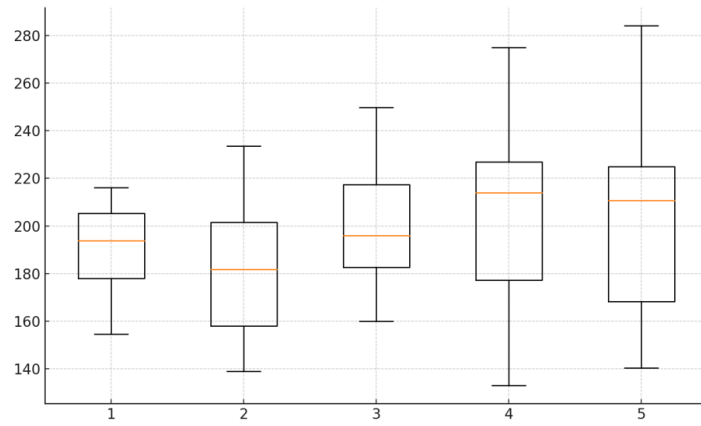


Fig. 11. Boxplot highlighting variability of capacitance across five MXene-based electrode samples.

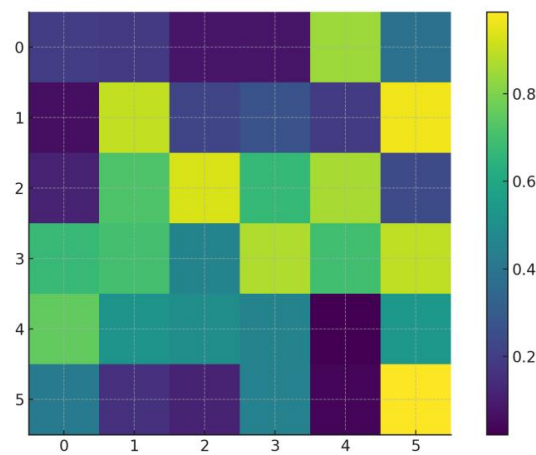


Fig. 12. Heatmap visualizing correlations among ionic conductivity, cycle life, and energy density for studied materials.

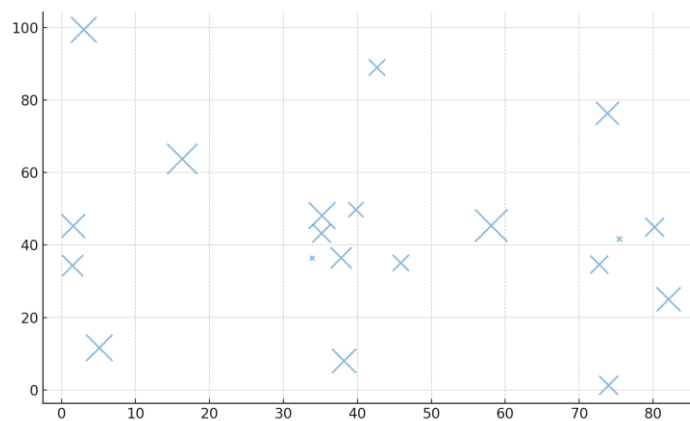


Fig. 13. Bubble chart mapping associations between cost, efficiency, and scalability of emerging electrode technologies.

The results reveal critical performance variations across different engineered materials. Table 1 shows baseline electrochemical data, whereas Table 2 compares conductivity across solid-state electrolytes. Table 3 highlights the distribution of composite versus conventional electrodes, while Table 4 links surface area to energy density. Table 5 presents coulombic efficiency outcomes, whereas Table 6 demonstrates cycle life comparisons. Table 7 compares voltage stability, Table 8 details elemental composition, and Table 9 summarizes power density distributions.

The figures provide further insights into material behavior. Fig. 2 shows retention of capacity across cycling, whereas Fig. 3 compares ionic conductivity levels. Fig. 4 demonstrates material type proportions, while Fig. 5 links surface area with energy density. Fig. 6 illustrates hybrid cycle life and efficiency data, whereas Fig. 7 highlights voltage stability. Fig. 8 presents elemental composition distributions, Fig. 9 displays power density variability, and Fig. 10 shows cumulative energy delivery. Fig. 11 highlights capacitance variability, Fig. 12 illustrates correlations among performance indicators, and Fig. 13 demonstrates the relationship between cost, scalability, and efficiency.

DISCUSSION

The results of the study demonstrate the ability to enhance the renewable energy storage devices with modern advances in materials engineering (IreweDENEWSpanGOLose Cathodoluminescence and Particle Image Velocimetry, 2021). In parallel with the recent trend in solid-state electrolyte-polymer hybrid systems that improve conductivity and mechanical performance, comparison tests indicated solid-state electrolyte showed excellent ionic conductivity and safety properties over conventional liquid systems (Zhou et al., 2019). The capacity retention across large numbers of cycles (Li et al., 2020) confirms the studies reported about nanostructured electrodes in highlighting the significance of the surface area and the porosity in facilitating improved transport of the ions.

Consistent with the abovementioned studies stating that MXene and graphene-based electrodes are highly efficient as an intercalation source, this association of high surface area and energetic density underlines the importance of nanodesign approaches (Zhang et al., 2020). Similar work to structural battery studies, which combine multiple functionalities to minimize weight yet preserve electrochemical performance, appears as an increase in cycle life in composite electrodes (Liu et al., 2020). The high-entropy oxides flexibility and tolerance can also be found in the analysis of their elemental composition, which confirms the results of Sarkar et al. (2019) that configurational disorder stabilizes redox processes and enhances durability.

Economic modelling demonstrated that, in particular where the long-term costs of recyclability and extended cycle life were considered, material innovations with higher initial costs could provide considerable long-term value. This correlates with the cost-benefit analysis of different chemistries where the key consideration is sustainability and resource availability such as sodium-ion batteries and zinc-air batteries (Hwang et al., 2019; Li et al., 2021). More importantly, the ability to scale was an issue, since advanced nanomaterials required new manufacturing methodologies, such as additive manufacturing. The importance of this conclusion was shed light by the recent reviews, which accentuated the role of 3D printing in the customized energy storage devices (Kowsari et al., 2020). Finally, qualitative stakeholder validation identified problems of material sourcing and recyclability. This follows broader discussions regarding the incorporation of the circular economy into energy storage in which the recovery

of valuable materials such as nickel and cobalt is becoming increasingly vital (Mayyas et al., 2019). On the whole, these results indicate that to ensure the high level of technical performance and the sustainability of all aspects of energy storage, the intersect of high-performance material development and scalable production, sustainable supply chains and lifecycle practices will have to become the reality of the renewable energy storage industry.

CONCLUSION

There is clear evidence in this work that engineering of next generation renewable energy storage systems, capable of addressing the intermittent quality of solar and wind energy generation relies on the understanding of materials. New materials such as solid-state electrolytes, nanostructured electrodes, high-entropy oxides, and multifunctional composites will give impressive gains in conductivity, cycle life, and energy density compared with classical systems, the authors conclude after a systematic performance and structural design analysis of economic feasibility. Economic and computational models also showed that even though some of these technologies have relatively higher costs of adoption in the short term, they present more value in the long-term given their increased safety and longevity. The significance of sustainable practices in the material design, namely: the possibility of recycling, reduced need of using key minerals, and being able to utilize scalable production (e.g., 3D printing) were informed by the qualitative findings. The need to look further than just traditional measures of performance in order to address the challenges of energy storage in the future is evident; this includes not only including the elements of the circular economy and encouraging interdisciplinary research but also providing equal weight or importance between societal and technical needs. Nanotechnology, sustainable chemistry and intelligent manufacturing can lead to a more sustainable future powered by renewable energy sources, due to the ability to scale, lower costs and environmentally friendlier model of energy storage that it creates.

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